

Convergence?

Stability of RNA hairpins

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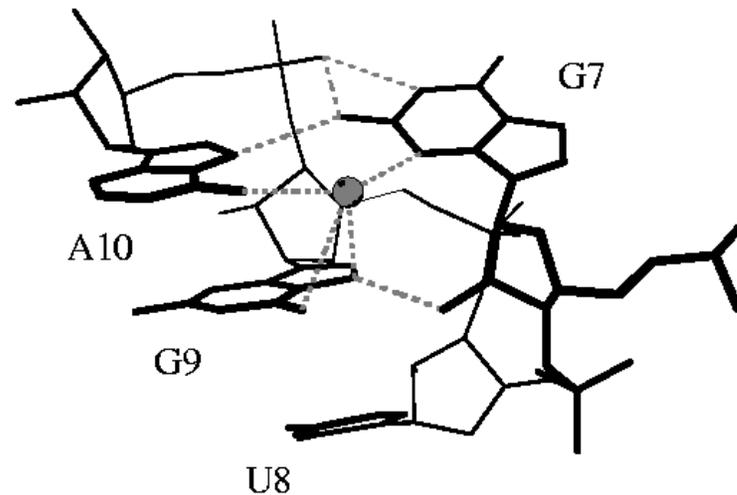
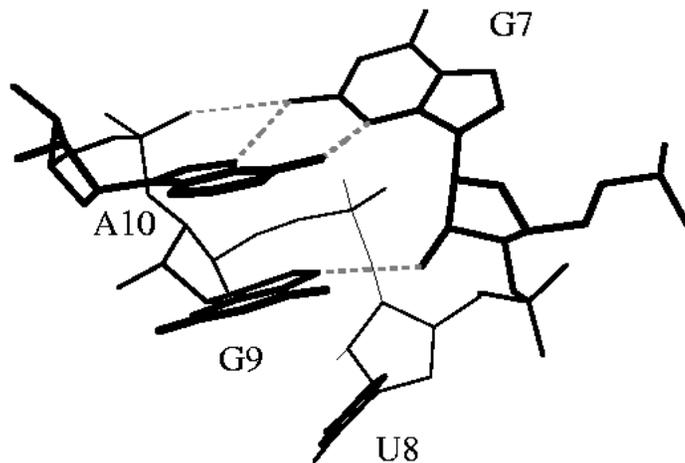
Lennart Nilsson, Karolinska Institutet

RNA Tetraloops – common motifs involved in tertiary contacts, protein binding, folding initiation

Of 256 possible tetraloops, 12 dominate in nature.

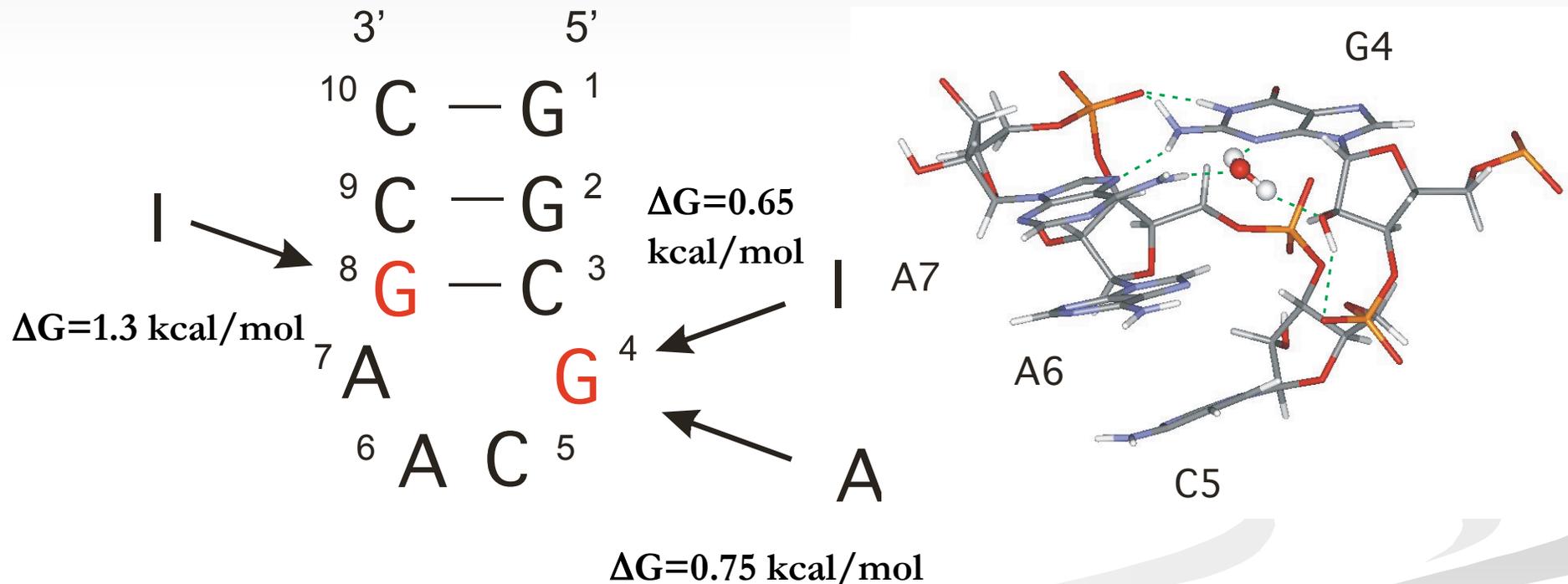
UNCG, GNRA, CUYG

~50% of rRNA tetraloops are GNRA loops:



Stability of a GNRA tetraloop

experimental destabilizing effects of point mutations

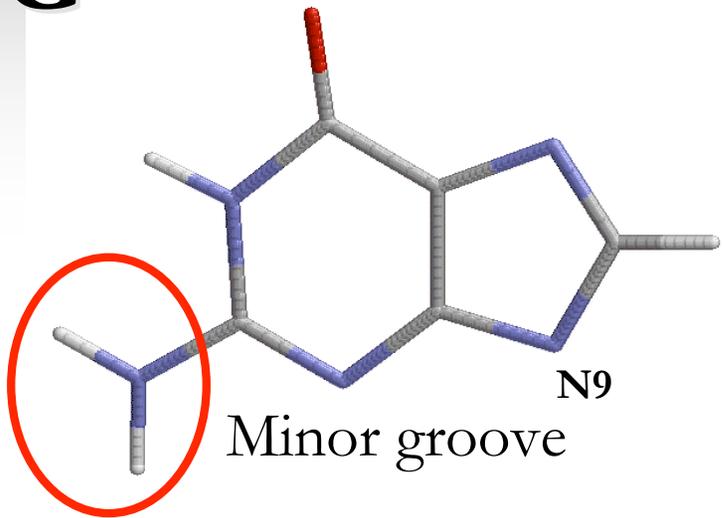


Thermodynamics: SantaLucia Jr., J., Kierzek, R. & Turner, D.H. (1992) *Science* **256**, 217-219.

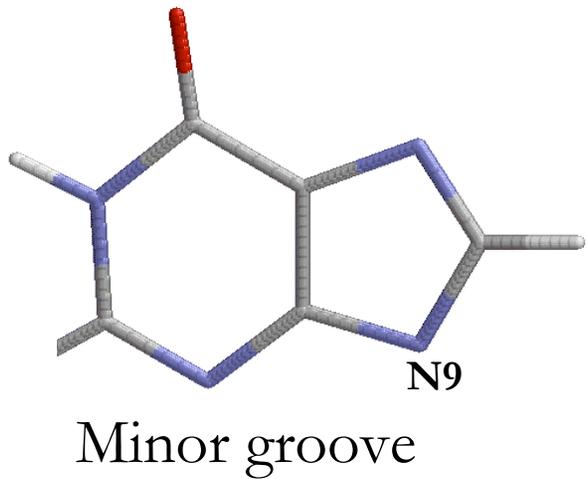
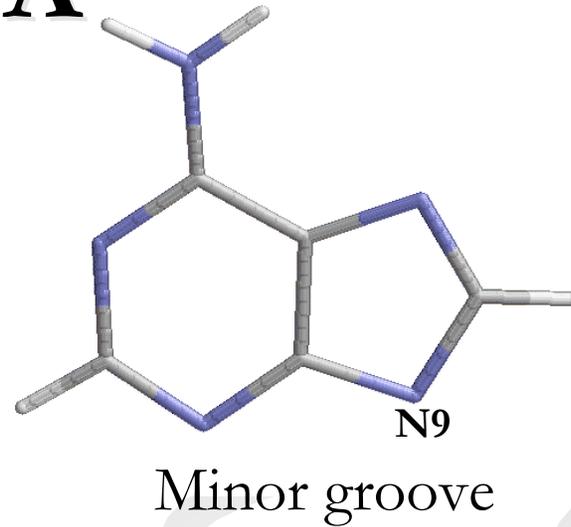
NMR (wt 1zih): Jucker FM, Heus HA, Yip PF, Moors EHM & Pardi A. (1996) *J. Mol. Biol.* **264**, 968-980.

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G



A



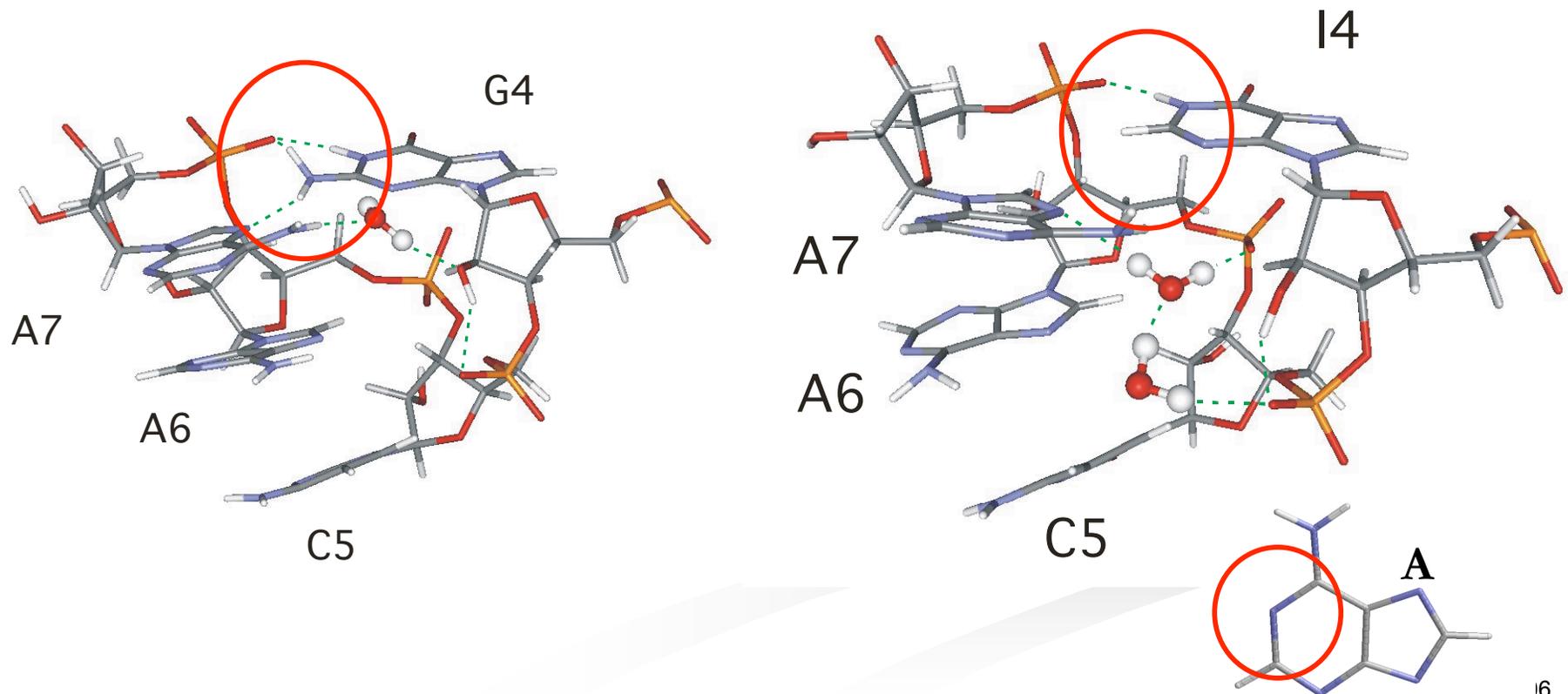
I

What is happening?

G → I: removal of NH₂ in minor groove

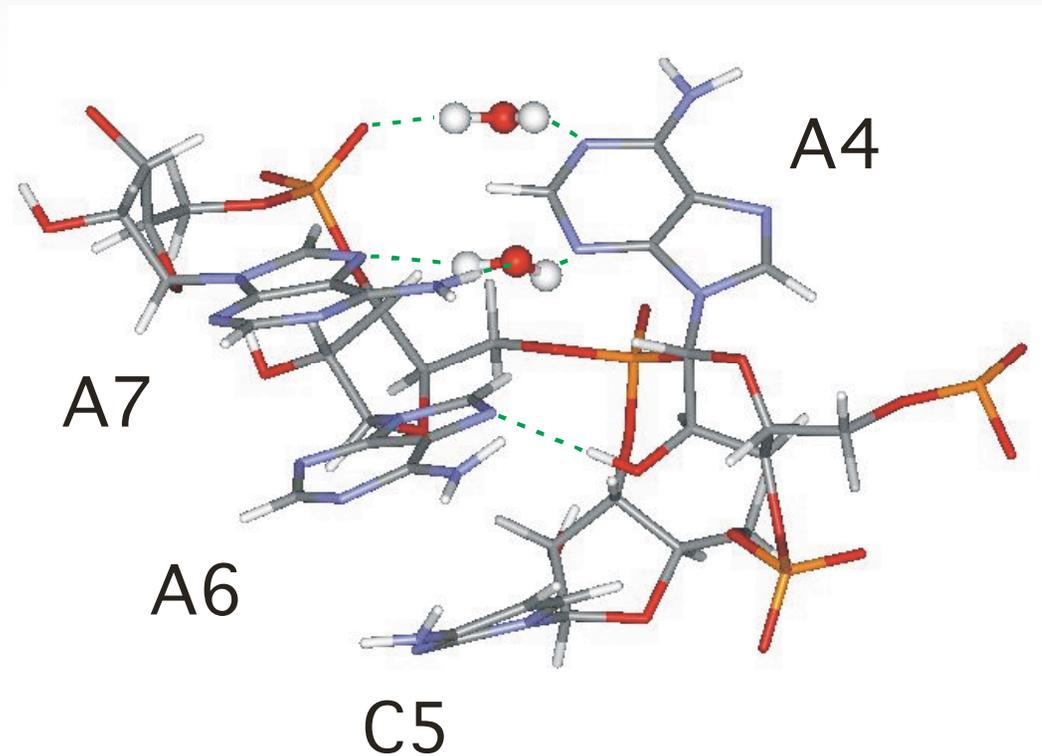
G → A: also loss of N-H at position 1

- G8 → I8 in the stem, loss of one H-bond (1.3 kcal/mol)
- G4 → I4 in the loop, loss of two H-bonds (0.65 kcal/mol)
- G4 → A4 in the loop, loss of three H-bonds (0.75 kcal/mol)

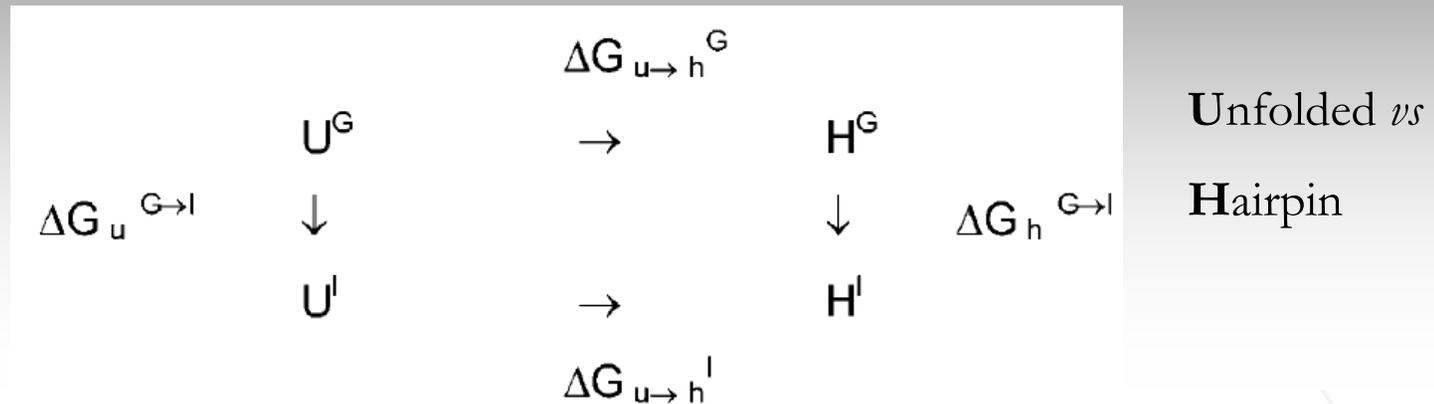


ACAA loop

no specific A4-A7 interactions



FEP simulations



Models of unfolded state

Average ΔG_u [kcal/mol]

	$G \rightarrow I$	$G \rightarrow A$
Single Guanosine	73.6 ± 0.1	74.5 ± 0.1
CGC triplet in A-RNA conformation	73.9 ± 0.2	74.3 ± 0.4
CGC triplet from the loop	74.4 ± 0.2	75.3 ± 0.1
GCGCA in A-RNA conformation	73.6 ± 0.1	

PERT, PBC w/ or w/o PME, spherical geometry, 101 windows, 0.4 – 0.8 ns total time

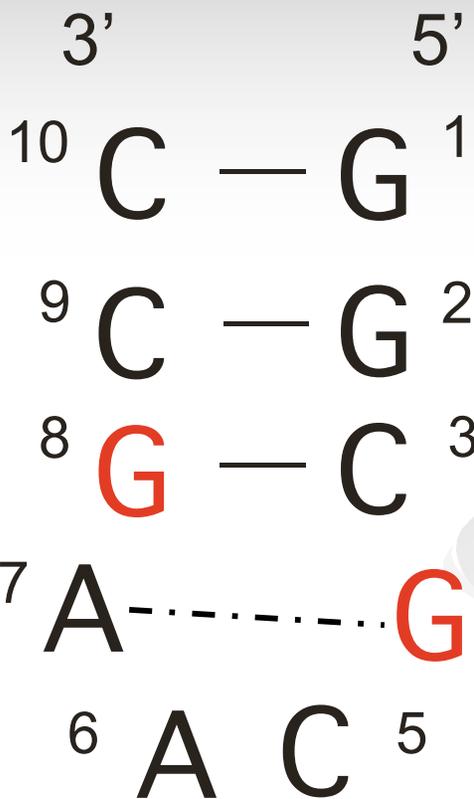
Sarzynska J, Nilsson L, Kulinski T. (2003). Effects of Base Substitutions in an RNA Hairpin from Molecular Dynamics and Free Energy Simulations. *Biophys. J.* **85**(6):3445-3459.

Calculated (PERT) hairpin stabilities

Sensational!

$$\Delta \Delta G_{\text{calc}} = 1.7 \pm 0.7$$

$$\Delta \Delta G_{\text{exp}} = 1.3$$



$$\Delta \Delta G_{\text{calc}} = 3.3 \pm 0.7$$

$$\Delta \Delta G_{\text{exp}} = 0.65$$

OK!



$$\Delta \Delta G_{\text{calc}} = 2.9 - 6.2$$

$$\Delta \Delta G_{\text{exp}} = 0.75$$

Auguste Comte, Philosophie Positive (1830):

“Every attempt to employ mathematical methods in the study of chemical questions must be considered profoundly irrational and contrary to the spirit of chemistry. If mathematical analysis should ever hold a prominent place in chemistry - an aberration which is happily almost impossible - it would occasion a rapid and widespread degeneration of that science.”

(mis?)calculated hairpin stabilities

TABLE 8 Free energy perturbation results from simulations in water sphere [kcal/mol]

Mutation	ΔG^* (forward/backward)	ΔG_{AVG}	$\Delta\Delta G_{\text{CALC}}$
Unfolded [†] G4→A4 [‡]	74.4/−74.2	74.3	
Hairpin G4→A4 [§]	79.6/−74.7	77.2	2.9
Hairpin G4→I4	77.0/−78.0	77.5	
Hairpin I4→A4	4.2/−1.8	3.0	
Hairpin G4→A4 via I4	81.2/−79.8	80.5	6.2
Hairpin I4→G4 [¶]	−74.0		
Hairpin A4→G4 via I4 [¶]	−75.8		4.2

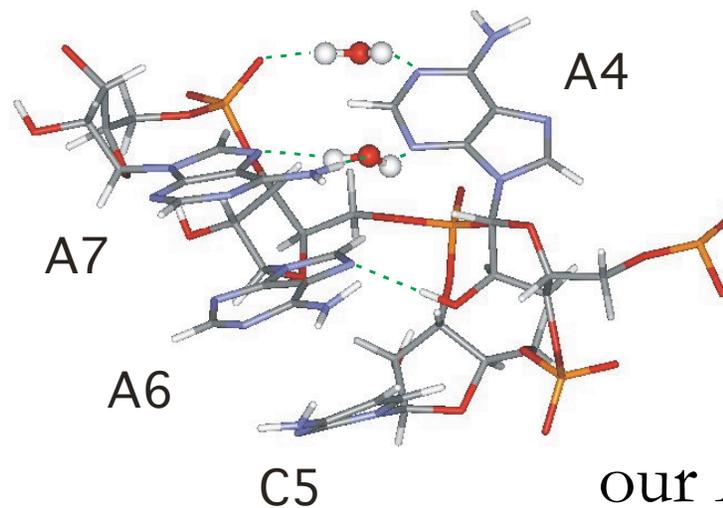
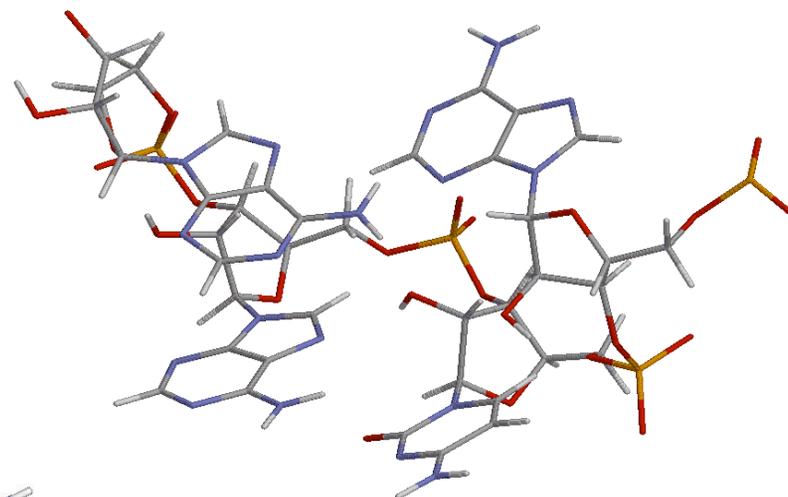
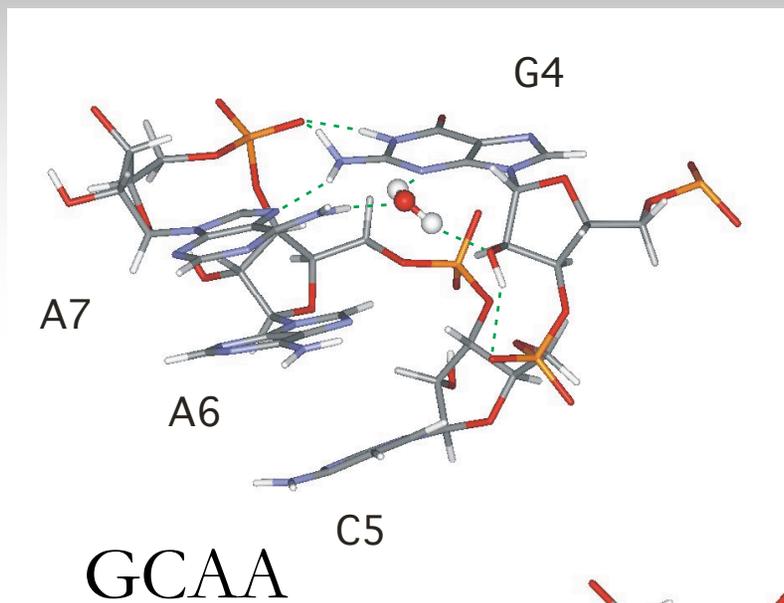
Conformational heterogeneity of the **A**CAA loop
(when imposed on a GCAA structure)

Conclusion

crucial for the rational design of RNA-derived molecules of therapeutic application. Our study suggests that GNRA tetraloops where G is replaced by I or A may have other low-energy conformations distinct from the GNRA fold, which were not reached during FEP simulations. This observation is consistent with the idea that the phylogenetic preference

Note added in proof: After this paper was accepted, the structure of an ACAA tetraloop was determined by NMR (Staple, D. W., S. E. Butcher. 2003. Solution structure of the HIV-1 frameshift inducing stem-loop RNA. *Nucl. Acids. Res.* 31:4326–4331) and found to be of the AGNN-tetraloop type, rather than of the GNRA-type. This confirms our suggestion that

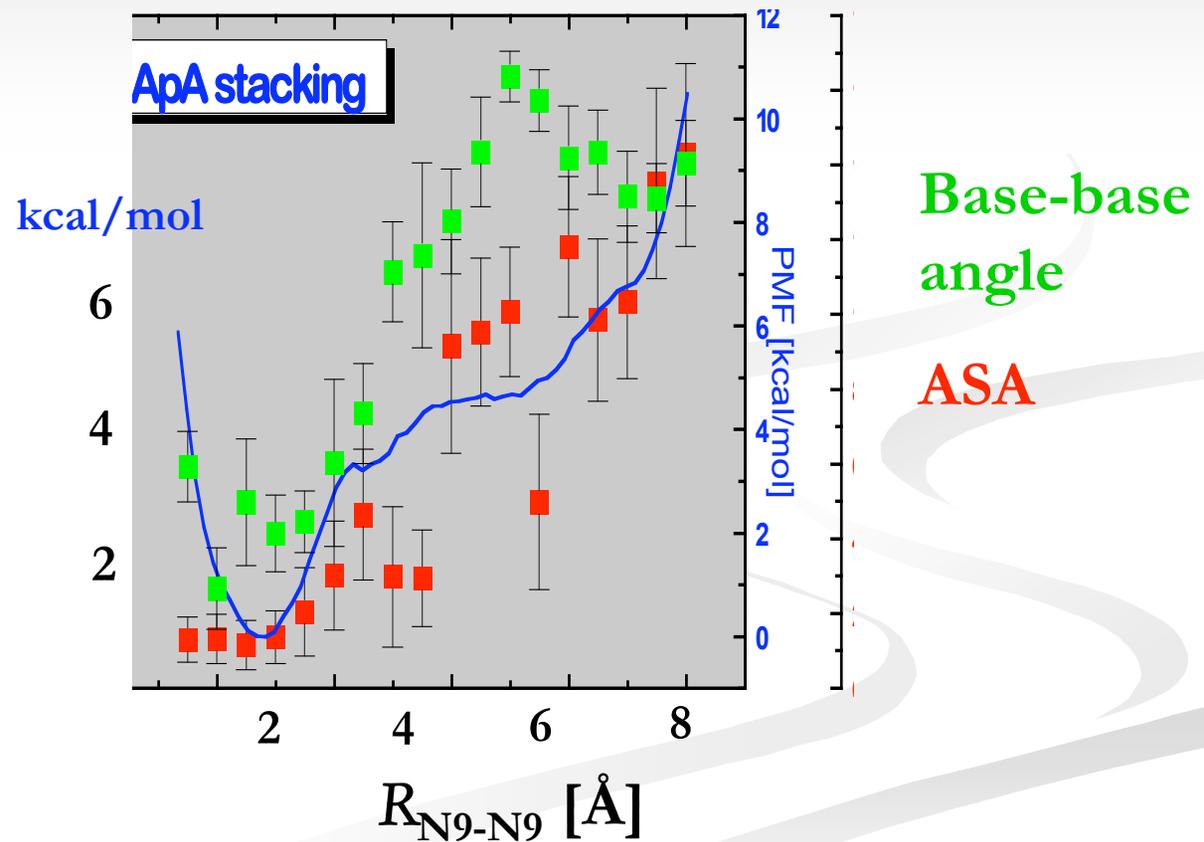
ACAA – sim vs real



Conformational Equilibria

Stacking Free Energies from PMF Calculations on Dinucleotides

- Sequence
- Length
- Temperature
- DNA / RNA
- Solvent



Temperature dependenc of ApA stacking from PMF *calculations*

13058 *J. Phys. Chem., Vol. 99, No. 35, 1995*

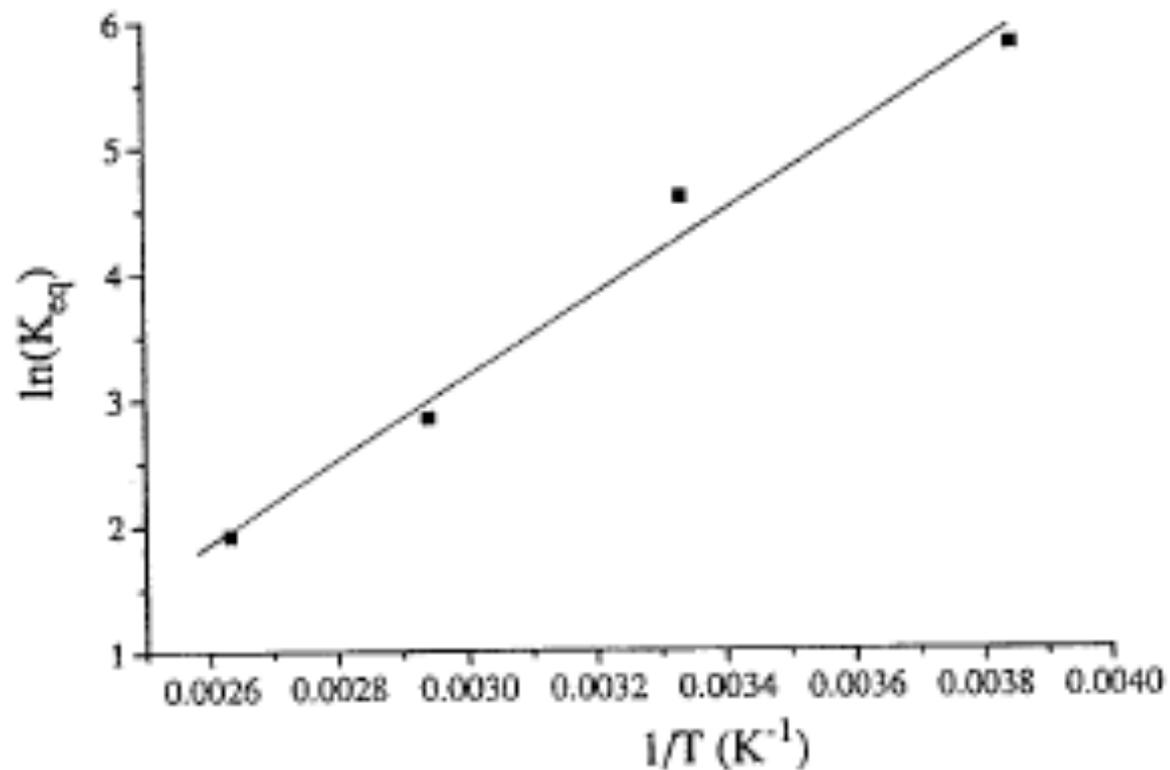


Figure 4. Van't Hoff plot for adenylyl-3',5'-adenosine. The line is

Conformational Equilibria

- Sequence

Pu-Pu > Pu/Py > Py-Py

- Temperature

Enthalpy driven

$\Delta H = -6 \text{ kcal/mol}$

$\Delta S = -13 \text{ cal/mol/K}$

- Primarily nearest neighbor effect

- High dielectric solvents favor stacking

- DNA/RNA

T better than U

2'OH - slight favor of stack

- Base mobility decreases stacking

Norberg&Nilsson, *Solvent Influence on Base Stacking*.

Biophys J., 1998. **74**(1): 394-402 (and references therein)